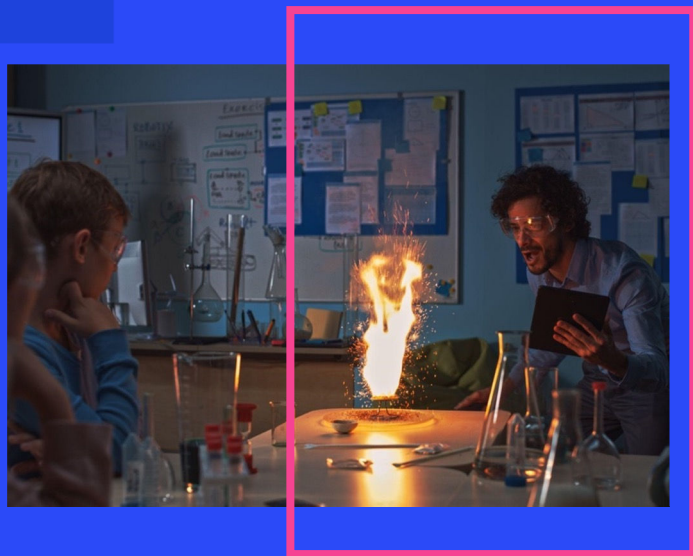




Keep Calm and Chemistry On: Successful Lab Activities for the New Chemistry Teacher

Laurie Nixon Watauga High School

- 25 years teaching
- Currently teaching AP[®] Chemistry and Honors Chemistry
- Carolina consultant for more than 20 years



Building Toward 3-Dimensional Learning

Science and Engineering Practices	Disciplinary Core Ideas	Crosscutting Concepts
▪ Developing and using models ▪ Planning and carrying out investigations ▪ Analyzing and interpreting data ▪ Constructing explanations	PS 1: Matter and its interactions PS 3: Energy	▪ Cause and effect: Mechanism and explanation ▪ Scale, proportion, and quantity ▪ Systems and system models ▪ Energy and matter: Flows, cycles, and conservation¹

1. NGSS Lead States, Next Generation Science Standards: For States, By States (Washington, DC: The National Academies Press, 2013), retrieved from www.nextgenscience.org or ngss.nsta.org

Workshop Safety



Workshop Reminders

- PowerPoint® for this workshop is available at knowledge.carolina.com. Type “workshop” in the search bar.
- More demos? Chemistry webinar available at knowledge.carolina.com. Type “webinar” in the search bar.
- Handout includes all demos and activities from the workshop.



Highly Visual Chemistry Phenomena for 3D Instruction - Web

44:30

Presenters: Matt Bostic and Chris Petersen, Carolina Product Developers

Grades: 6-12

What are the aspects of good phenomena? How can you redesign classic, tried-and-true chemistry demos to support 3-dimensional learning? Find out in this webinar.

Demo: Frozen Beaker

An extreme endothermic reaction

Goggles On! Carolina Chemonstration® in progress

Solid barium hydroxide octahydrate and solid ammonium chloride

Predict

Observe

Explain



This reaction may reach a temperature as low as -30°C .

Demo: Frozen Beaker

An extreme endothermic reaction

Curriculum connections:

- Laws of thermodynamics
- Energy changes
- Endothermic vs. exothermic reactions

Possible discussion questions:

- What would be some practical, real-world applications of an endothermic reaction?
- Could you measure the heat of reaction in this demo? If not, why?
- Are heat and temperature the same thing?



Lab: Mystery Chemical Reactions

- Explore scientific phenomena
- Identify visible signs of reaction (precipitate, gas, and/or color change)
- Microscale chemistry benefits (save time and money; reduce waste)

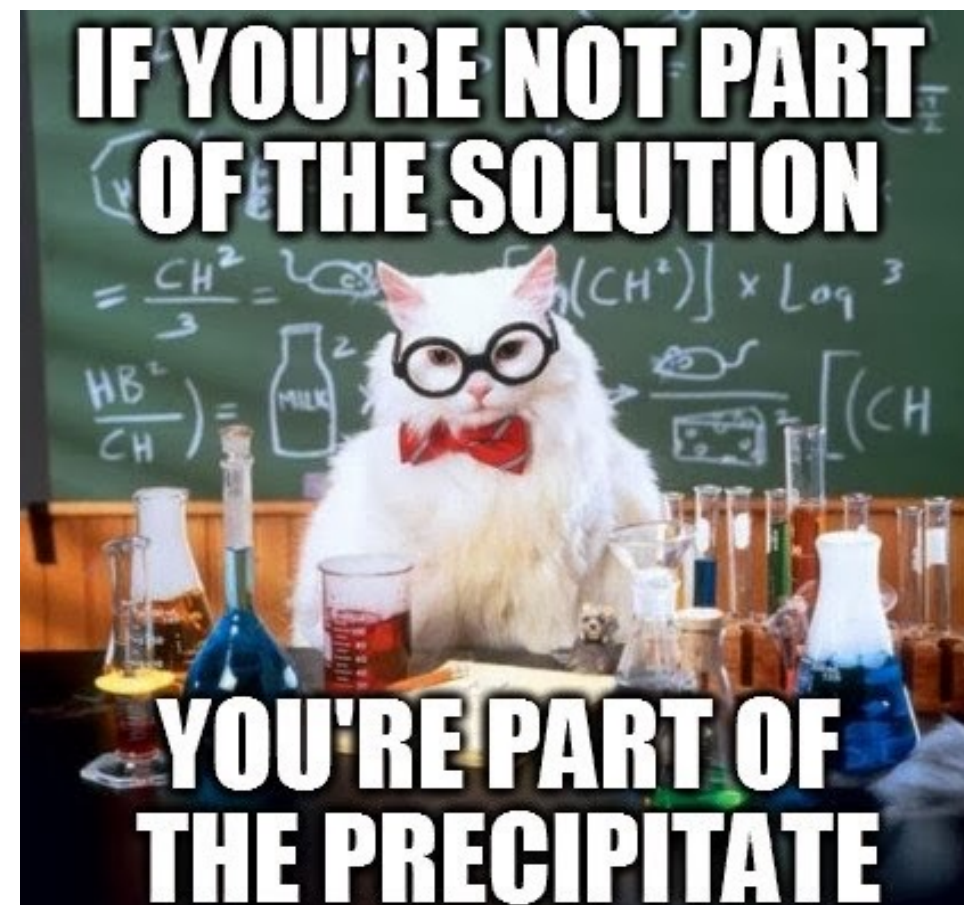
Featured Digital Content

- Interactive lessons



Lab: Mystery Chemical Reactions

	sodium phosphate	iron(III) chloride	copper(II) sulfate	potassium iodide	lead(II) nitrate	sodium carbonate	silver nitrate	calcium chloride	sodium hydroxide
hydrochloric acid		yellow precipitate	light blue precipitate				white precipitate		
sodium hydroxide		yellow precipitate	light blue precipitate				brown precipitate		
calcium chloride		yellow precipitate	light blue precipitate				white precipitate		
silver nitrate		yellow precipitate	light blue precipitate	yellow precipitate					
sodium carbonate		yellow precipitate	light blue precipitate		white precipitate				
lead(II) nitrate		yellow precipitate	light blue precipitate	yellow precipitate					
potassium iodide		red precipitate	light blue precipitate						
copper(II) sulfate		orange precipitate							
iron(III) chloride		yellow precipitate							

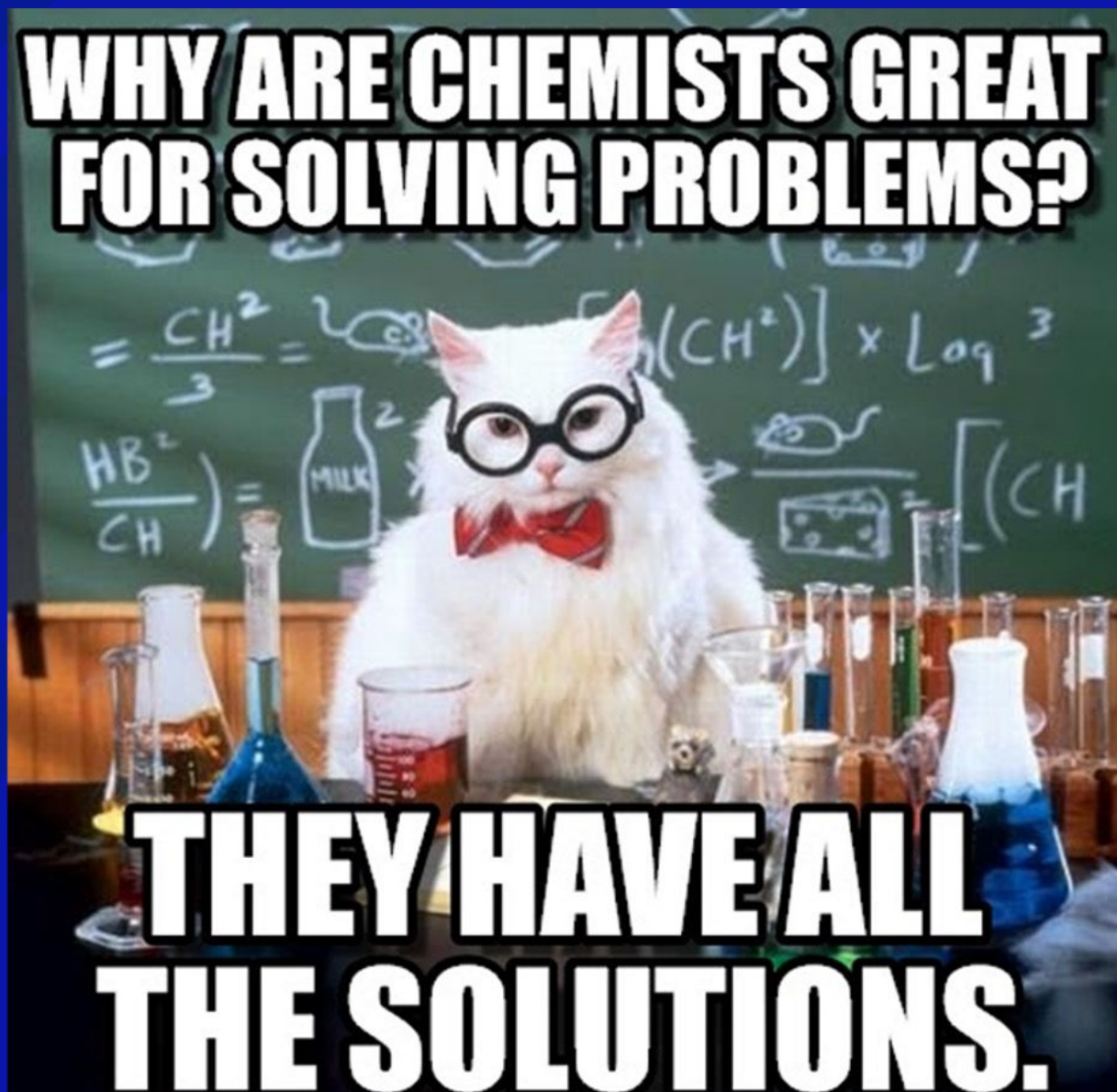


Lab: Balancing Chemical Equations

A tactile introduction to stoichiometry

- Understand the Law of Conservation of Mass
- Understand the difference between coefficients and subscripts in chemical equations





Workshop Kit Review



**Endothermic Reactions:
Beaker Freezer**

Item #840378



Mystery Chemical Reactions

Item #840660



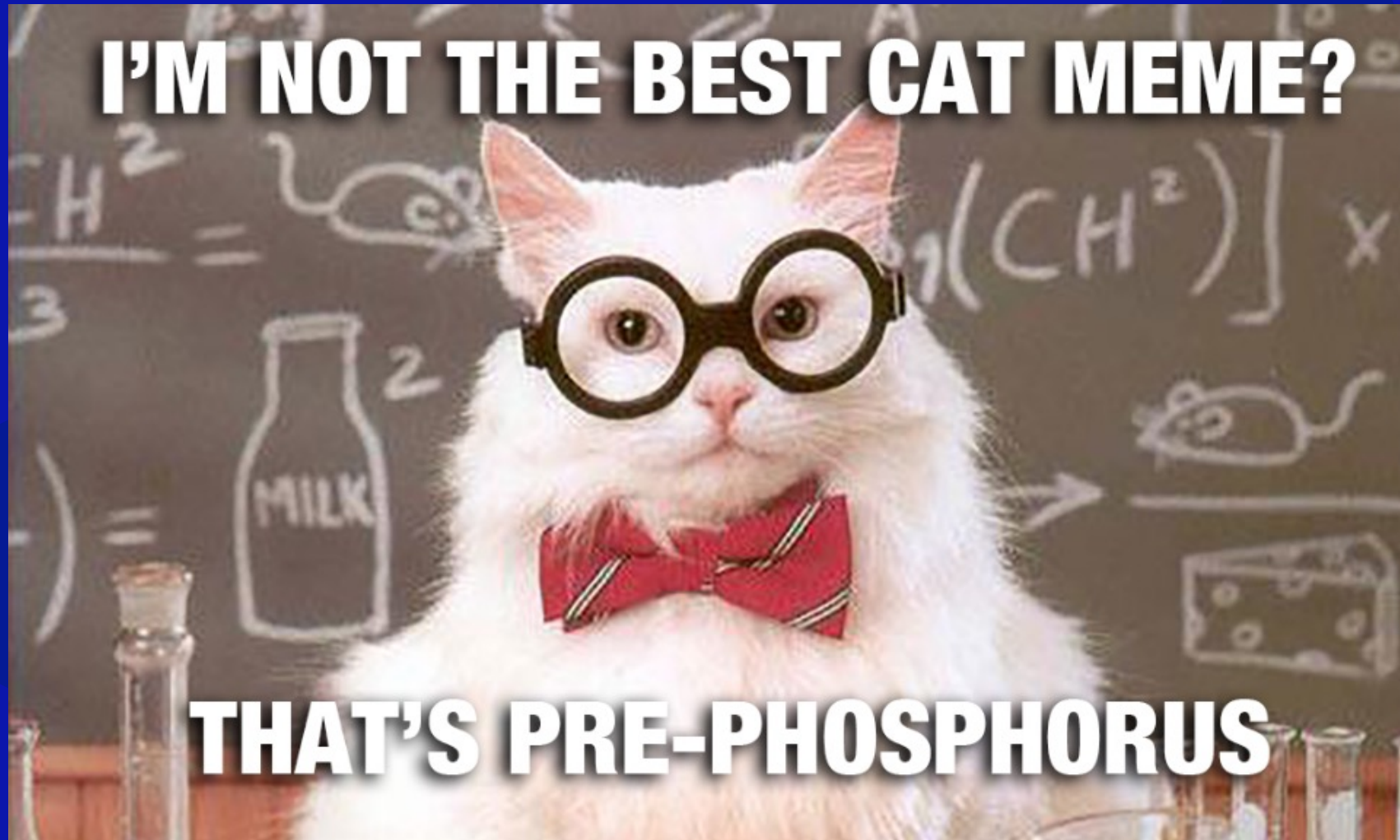
Petri Dish Electrolysis

Item #840830



Balancing Chemical Equations

Item #840656



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