



Keep Calm and Chemistry On: Successful Lab Activities for the New Chemistry Teacher

Laurie Nixon Watauga High School

- 25 years teaching
- Currently teaching AP[®] Chemistry and Honors Chemistry
- Carolina consultant for more than 20 years



Building Toward TEKS Science Concepts

Beaker Freezer

- **C.13.A:** Explain everyday examples that illustrate the four laws of thermodynamics.
- **C.13.C:** Classify processes as exothermic or endothermic and represent energy changes that occur in chemical reactions using thermochemical equations or graphical analysis.

Mystery Chemical Reactions

- **C.5.B:** Predict the properties of elements in chemical families, including alkali metals, alkaline earth metals, halogens, noble gases, and transition metals, based on valence electrons patterns using the Periodic Table.
- **C.5.C:** Analyze and interpret elemental data, including atomic radius, atomic mass, electronegativity, ionization energy, and reactivity to identify periodic trends.
- **C.9.A:** Interpret, write, and balance chemical equations, including synthesis, decomposition, single replacement, double replacement, and combustion reactions using the law of conservation of mass.
- **C.11.D:** Investigate the general rules regarding solubility and predict the solubility of the products of a double replacement reaction.

Building Toward TEKS Science Concepts

Balancing Chemical Equations

- **C.9.A:** Interpret, write, and balance chemical equations, including synthesis, decomposition, single replacement, double replacement, and combustion reactions using the law of conservation of mass.

Petri Dish Electrolysis

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Workshop Safety



Workshop Reminders

- PowerPoint® for this workshop is available at knowledge.carolina.com. Type “workshop” in the search bar.
- More demos? Chemistry webinar available at knowledge.carolina.com. Type “webinar” in the search bar.
- Handout includes all demos and activities from the workshop.



Highly Visual Chemistry Phenomena for 3D Instruction - Web

44:30

Presenters: Matt Bostic and Chris Petersen, Carolina Product Developers

Grades: 6-12

What are the aspects of good phenomena? How can you redesign classic, tried-and-true chemistry demos to support 3-dimensional learning? Find out in this webinar.

Demo: Frozen Beaker

An extreme endothermic reaction

Goggles On! Carolina Chemonstration® in progress

Solid barium hydroxide octahydrate and solid ammonium chloride

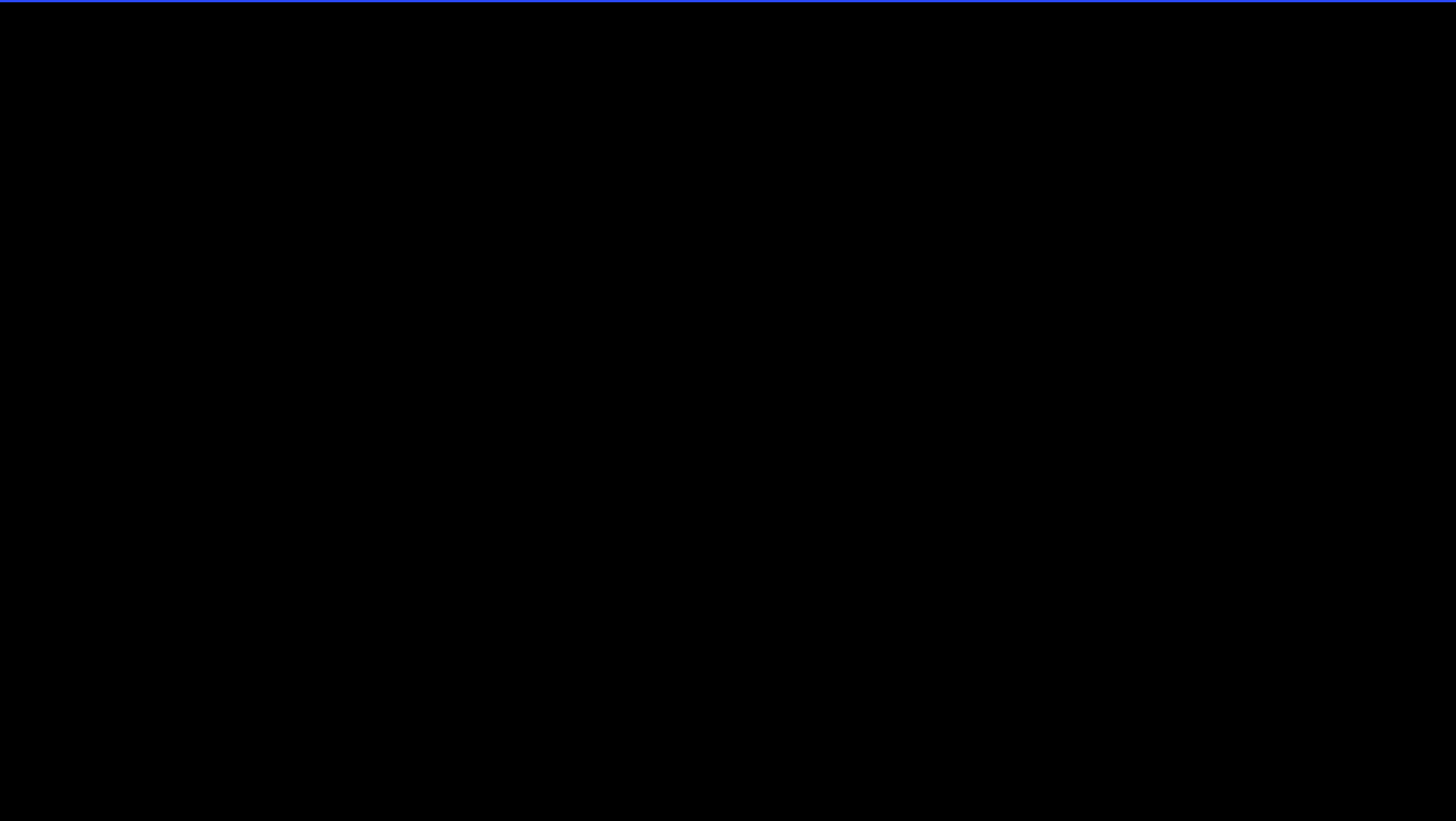
Predict

Observe

Explain



This reaction may reach a temperature as low as -30°C .



Demo: Frozen Beaker

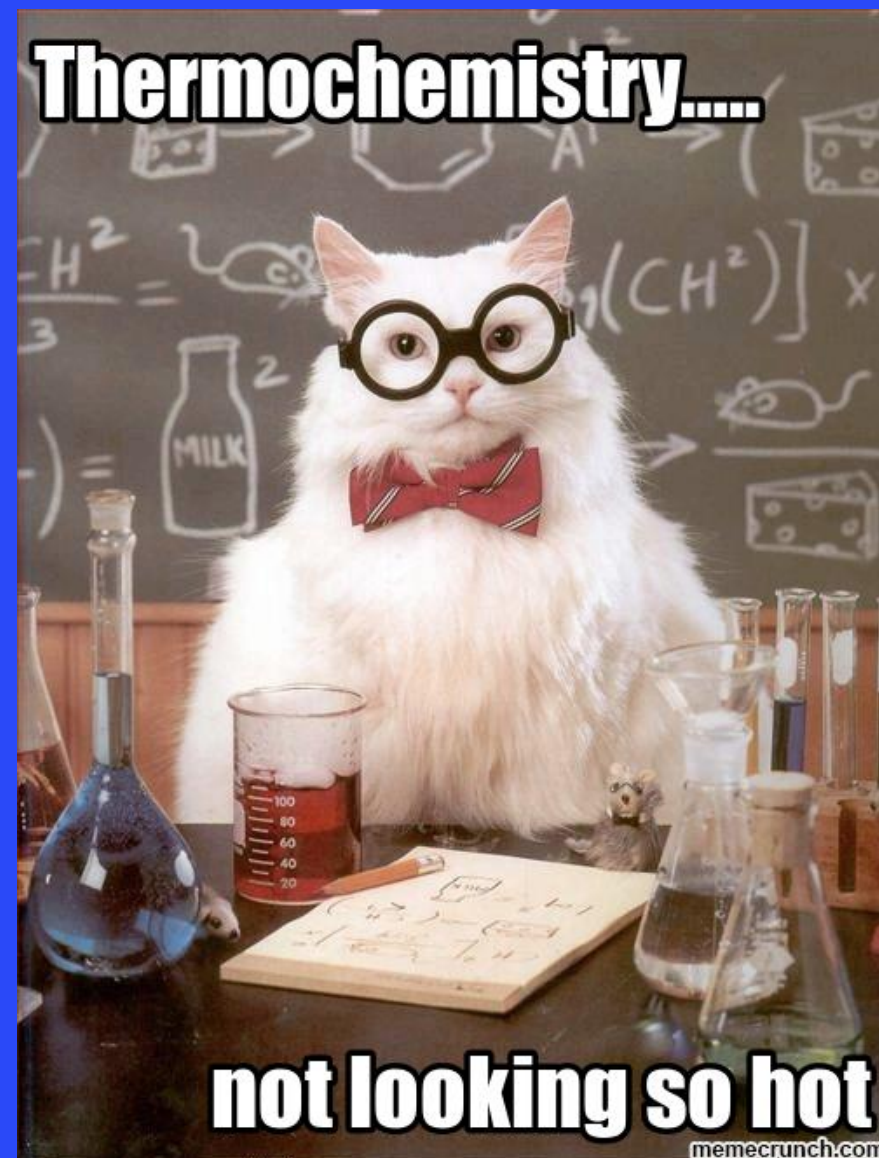
An extreme endothermic reaction

Curriculum connections:

- Laws of thermodynamics
- Energy changes
- Endothermic vs. exothermic reactions

Possible discussion questions:

- What would be some practical, real-world applications of an endothermic reaction?
- Could you measure the heat of reaction in this demo? If not, why?
- Are heat and temperature the same thing?



Lab: Mystery Chemical Reactions

- Explore scientific phenomena
- Identify visible signs of reaction (precipitate, gas, and/or color change)
- Microscale chemistry benefits (save time and money; reduce waste)

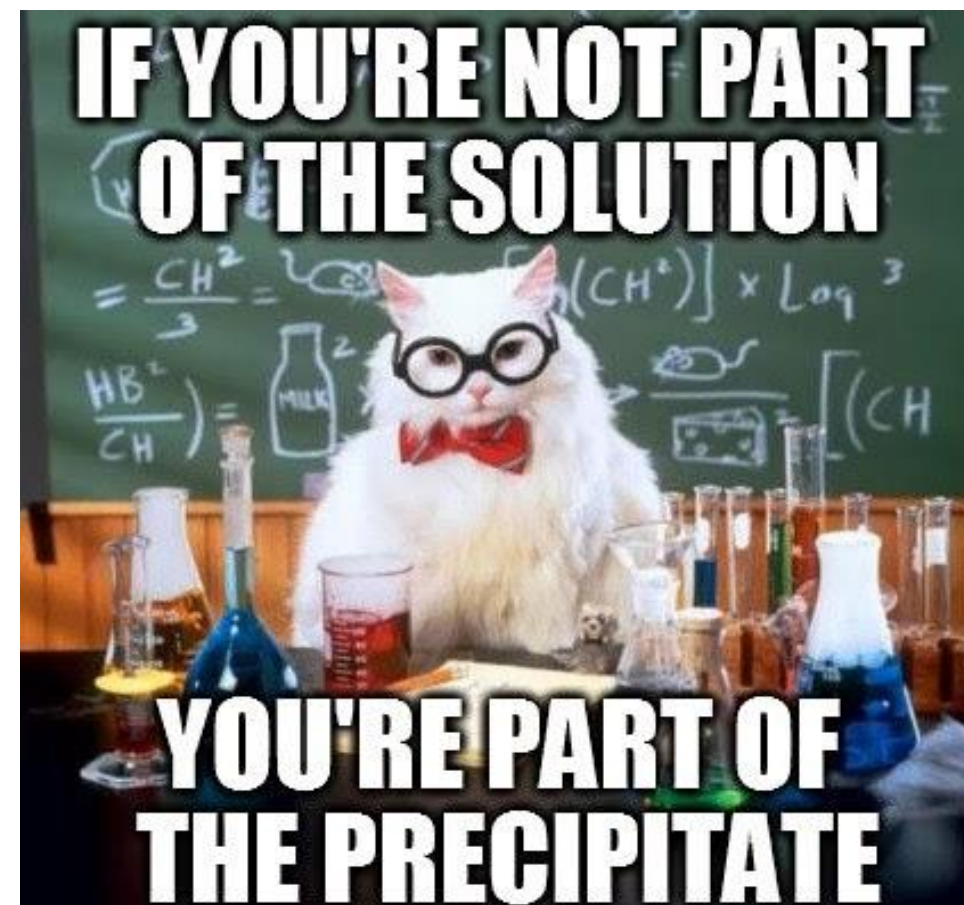
Featured Digital Content

- Interactive lessons



Lab: Mystery Chemical Reactions

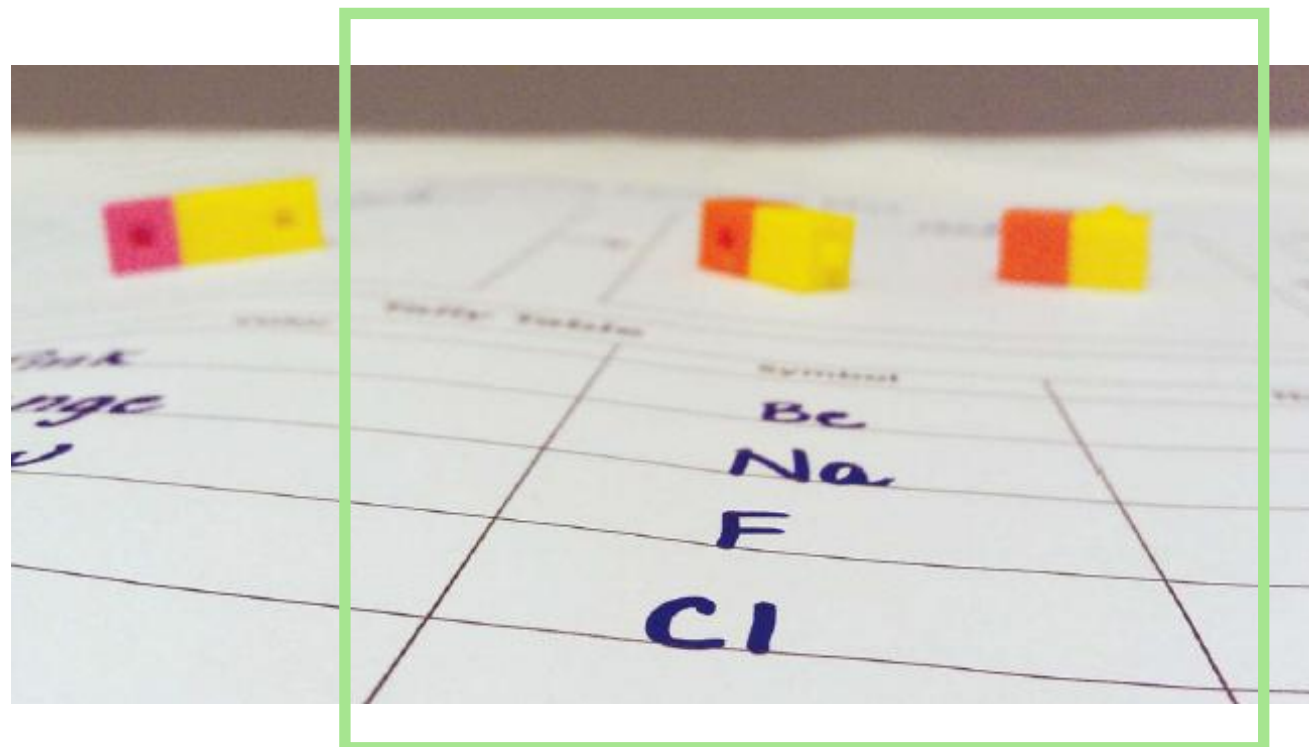
	sodium phosphate	iron(III) chloride	copper(II) sulfate	potassium iodide	lead(II) nitrate	sodium carbonate	silver nitrate	calcium chloride	sodium hydroxide
hydrochloric acid		yellow precipitate	light blue precipitate				white precipitate		
sodium hydroxide		yellow precipitate	light blue precipitate				brown precipitate		
calcium chloride		yellow precipitate	light blue precipitate				white precipitate		
silver nitrate		yellow precipitate	light blue precipitate	yellow precipitate					
sodium carbonate		yellow precipitate	light blue precipitate		white precipitate				
lead(II) nitrate		yellow precipitate	light blue precipitate	yellow precipitate					
potassium iodide		red precipitate	light blue precipitate						
copper(II) sulfate		orange precipitate							
iron(III) chloride		yellow precipitate							

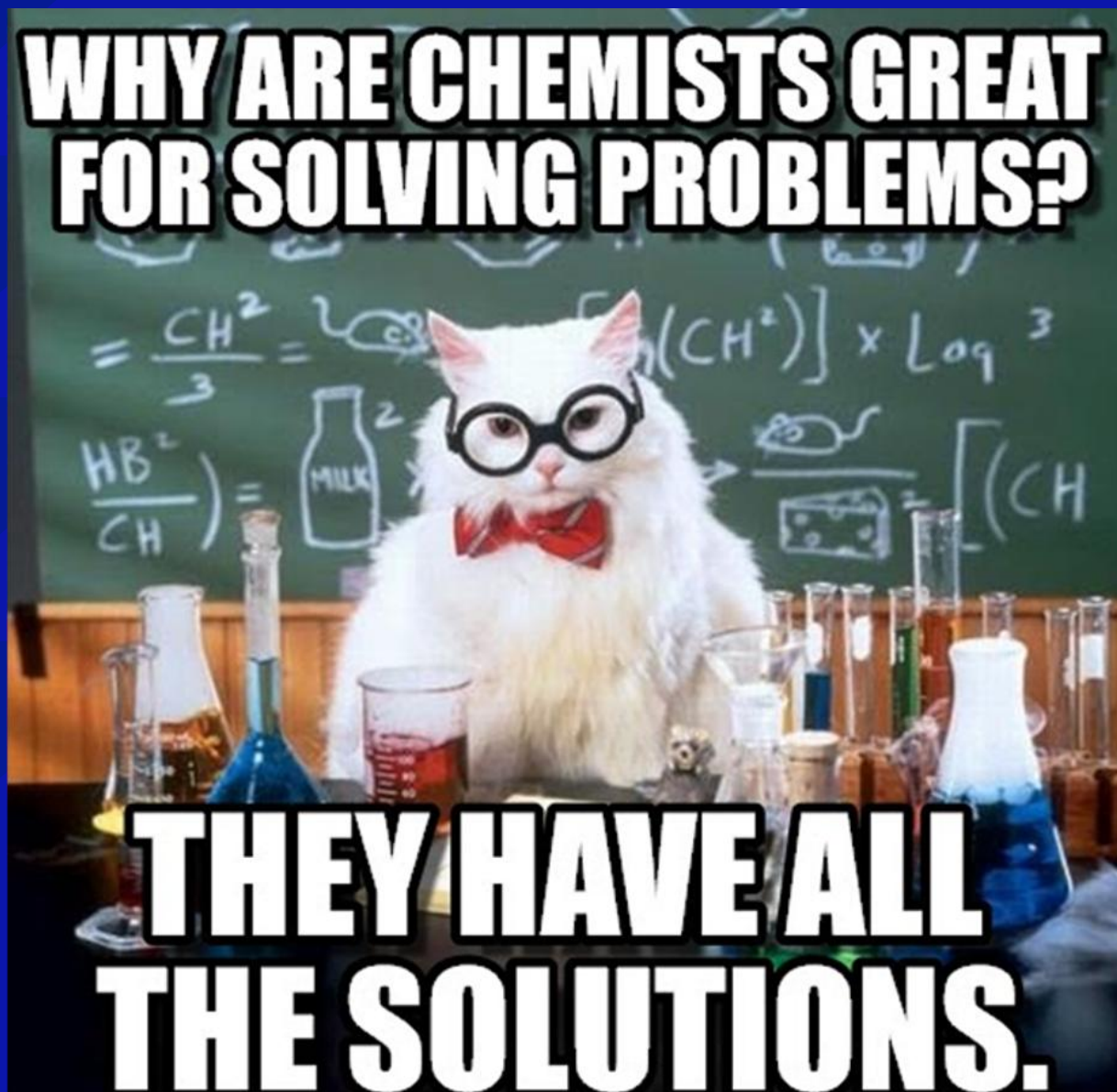


Lab: Balancing Chemical Equations

A tactile introduction to stoichiometry

- Understand the Law of Conservation of Mass
- Understand the difference between coefficients and subscripts in chemical equations





Workshop Kit Review



**Endothermic
Reactions:
Beaker Freezer**
Item #840378



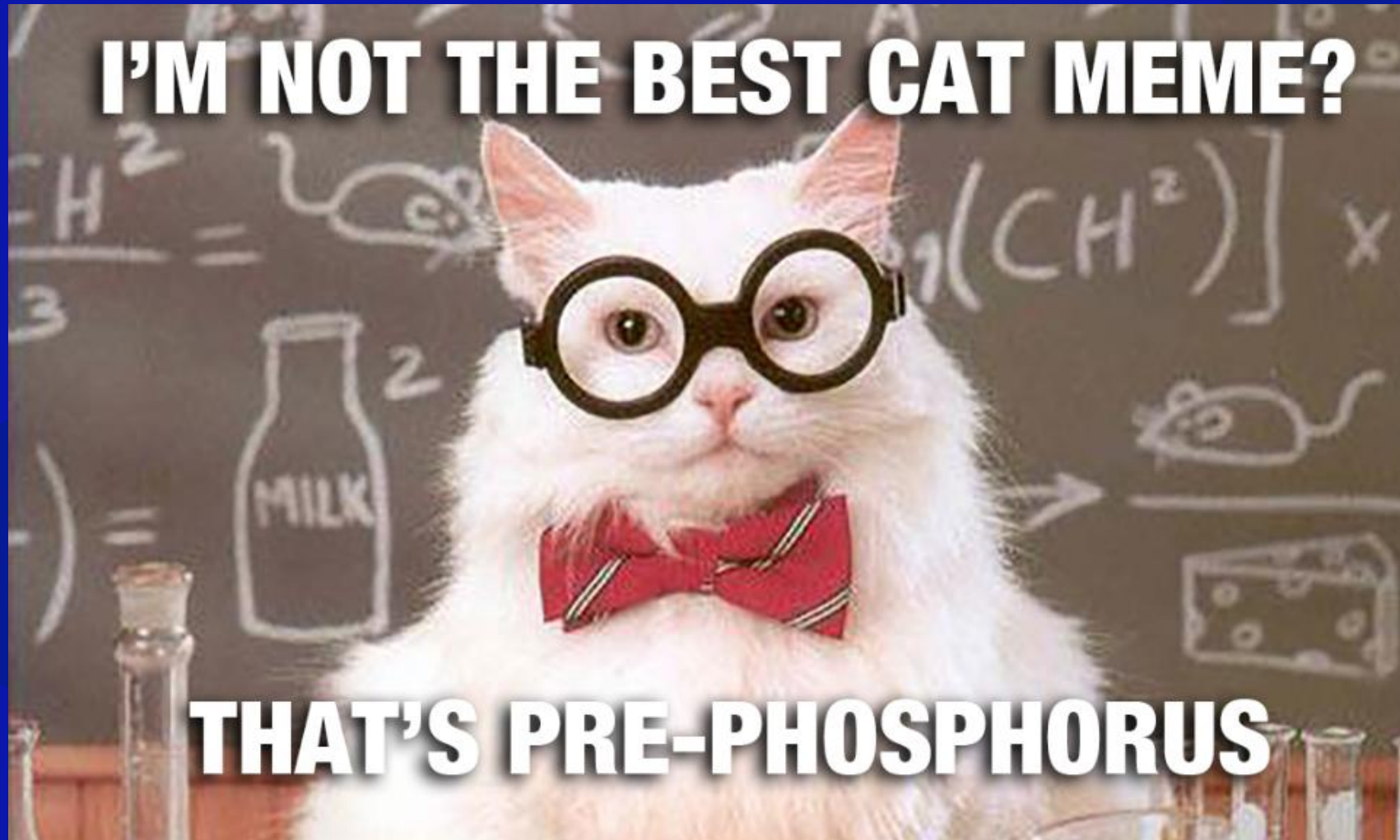
**Mystery Chemical
Reactions**
Item #840660



Petri Dish Electrolysis
Item #840830



**Balancing Chemical
Equations**
Item #840656



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